# Multitarget-Directed Benzylideneindanone Derivatives: Anti-β-Amyloid (Aβ) Aggregation, Antioxidant, Metal Chelation, and Monoamine Oxidase B (MAO-B) Inhibition Properties against Alzheimer's Disease

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Ⓢ *Supporting Information*

ABSTRACT: A novel series of benzylideneindanone derivatives were designed, synthesized, and evaluated as multitarget-directed ligands against Alzheimer's disease. The in vitro studies showed that most of the molecules exhibited a significant ability to inhibit self-induced β-amyloid ($A\beta_{1-42}$) aggregation (10.5−80.1%, 20 μM) and MAO-B activity (IC_{50} of 7.5−40.5 μM), to act as potential antioxidants (ORAC-FL value of 2.75−9.37), and to function as metal chelators. In particular, compound **41** had the greatest ability to inhibit $A\beta_{1-42}$ aggregation (80.1%), and MAO-B (IC_{50} = 7.5 μM) was also an excellent antioxidant and metal chelator. Moreover, it is capable of inhibiting Cu(II)-induced $A\beta_{1-42}$ aggregation and disassembling the well-structured Aβ fibrils. These results indicated that compound **41** is an excellent multifunctional agent for the treatment of AD.

Anti-Aβ aggregation: 80.1% at 20 μM
ORAC-FL value of 5.60 (Trolox equiv)
MAO-B IC_{50}: 7.50μM
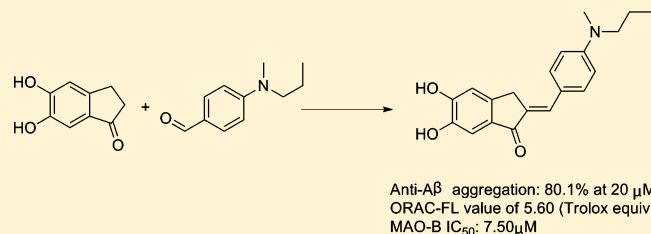
INTRODUCTION

Alzheimer' disease (AD) is a multifaceted, progressive neurodegenerative disorder characterized by progressive cognitive decline and memory loss. Although the etiology of AD is not fully known at present, several conditions have been considered to play significant roles in the pathogenesis of AD. These include the development of deposits of β-amyloid and τ-protein, oxidative stress, dyshomeostasis of biometals, and low levels of acetylcholine (ACh).[1] In particular, the production and accumulation of oligomeric aggregates of Aβ in the brain are a central event in the pathogenesis of AD according to the "amyloid hypothesis", as they are thought to be able to initiate the pathogenic cascade, ultimately leading to neuronal loss and dementia.[2] Aβ can efficiently generate reactive oxygen species in the presence of some transition metals and form stable dityrosine cross-linked dimers, which are generated by free radical attack under oxidative conditions.[3] Oxidative damage is present within the brain of AD patients, and it was shown to affect every class of biological macromolecules, including nucleic acids, proteins, lipids, and carbohydrates.[4] It is also an event that precedes the appearance of other pathological hallmarks of the disease, such as amyloid plaques and neurofibrillary tangles.[5,6] Therefore, antioxidant protection is important during aging and especially in AD patients as the endogenous antioxidant protection system declines rapidly.

Monoamine oxidases A and B (MAO-A and MAO-B) are important FAD-dependent enzymes (flavoenzymes) responsible for the metabolism of neurotransmitters such as dopamine, serotonin, adrenaline, and noradrenaline and for the inactivation of exogenous arylalkylamines.[7] Thus, the inhibition of MAO-A/MAO-B increases, primarily in the CNS, the levels of such neurotransmitters, which are decreased in AD patients compared to age-matched controls.[8] Selegiline, a selective MAO-B inhibitor, has been shown to significantly improve learning and memory deficits in animal models associated with AD and to slow the disease progression in AD patients.[9] This evidence suggested that selective MAO-B inhibitors seem to be an important treatment of AD.

Recently, transition metals, including copper and zinc, have been found to directly bind to amyloid plaques by spectroscopic studies.[10,11] Moreover, studies suggested that redox-active metal ions, such as Cu^{2+} and Fe^{2+}, are involved in the production of reactive oxygen species (ROS) and oxidative stress,[12] which implies that these biometals also play a central role in many critical aspects of AD. Thus, modulation of such biometals in the brain has been proposed as a potential therapeutic strategy for the treatment of AD.[13]

The multifaceted conditions of the AD state have encouraged active research in the development of multitarget-directed ligands (MTDLs) to act as agents for the treatment of this disease.[14−17] These drugs, which possess two or more complementary biological activities, may represent an important clinical advance in the future.
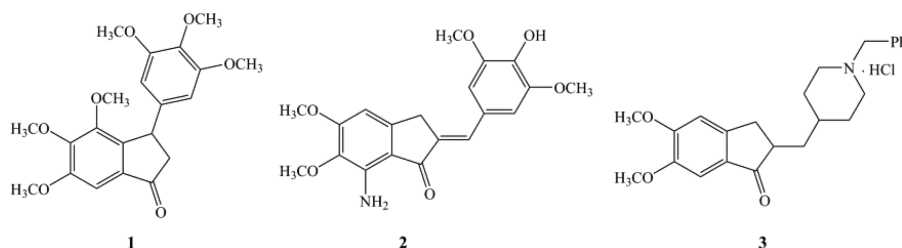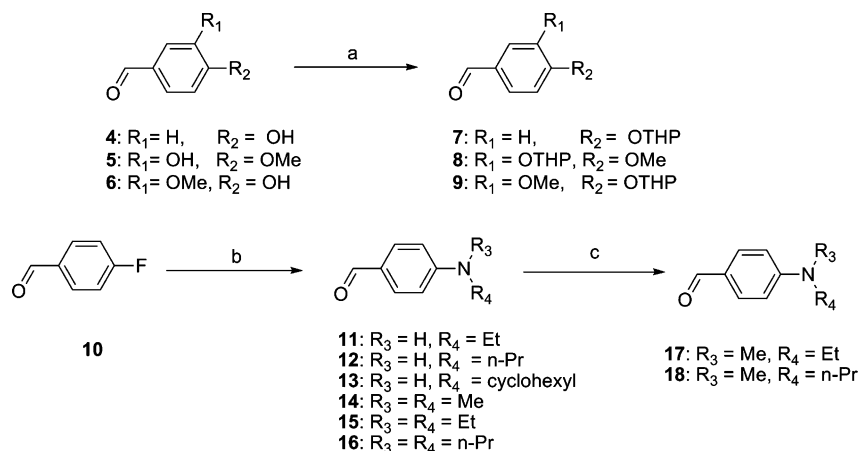
Figure 1. Some indanone derivatives with important bioactivities.

Scheme 1. Synthesis of the Aromatic Aldehyde Intermediates 7−9 and 11−18[a]

[a]Reagents and conditions: (a) 3,4-2H-dihydropyran, PPTS, DCM, rt; (b) R_3NR_4, TBAB, K_2CO_3, DMSO, 90°C; (c) 37% HCHO, HCOOH, reflux.

Indanones and their derivatives are important bioactive molecules that have been studied to determine their biological activities within disease states, including Alzheimer's disease and cancer. For example, gallic acid based indanone (**1**, Figure 1) derivatives and some imidazolyl-substituted 2-benzylide-neindanone derivatives were recently studied as inhibitors for cancer treatment.[18] Indanocine (**2**, Figure 1) and its analogues were developed to combat drug-resistant malignancies.[19] Donepezil hydrochloride (**3**, Figure 1), which acts as an AChE inhibitor, was approved by the FDA for the treatment of mild to moderate Alzheimer's disease.[20]

Previously, we reported the design, synthesis, and evaluation of 9-O and 9-N substituted berberine derivatives as dual or multifunctional agents for the treatment of Alzheimer's disease.[21,22] In this paper, we report the study of the design, synthesis, and evaluation of a novel series of indanone derivatives that were found to show potentially applicable biological activities, including the inhibition of self-induced $A\beta$ aggregation, inhibition of MAO-B activity, antioxidant properties, and metal chelation.

RESULTS AND DISCUSSION

Chemistry. The synthetic route of key intermediates **7−9** and **11−18** is shown in Scheme 1. The protection of hydroxy groups at the phenyl rings of **4−6** provided the THP-protected aromatic aldehyde derivatives **7−9**. The substitution of the fluorine atom of 4-fluoro benzaldehyde with a series of amines in the presence of TBAB gave intermediates **11−16**. In compounds **17** and **18**, the aldehydes with a substituted amino group at the 4-position were converted to **11** and **12** by aminomethylation with formaldehyde and formic acid in reflux.

According to Scheme 2, **20** was synthesized by the reaction of 5,6-dimethoxy-1-indanone with boron tribromide in

dichloromethane, which then reacted with iodomethane in the presence of lithium carbonate to provide 6-hydroxy-5-methoxyl-1-indanone **21**. The reaction of compound **21** with 3,4-2H-dihydropyran afforded the THP-protected indanone derivative **22**. The target compounds **23** and **24**, without the hydroxy group on indanone or aldehyde segment, were obtained by direct condensation of the commercially available 5,6-dimethoxy-1-indanone (**19**) and the appropriate aldehydes in ethanolic KOH solution. Condensation of 5,6-dihydroxy-1-indanone (**20**) and the corresponding amine, catalyzed by TsOH, gave compounds **38−41**. Finally, compounds **25−37** were obtained by the condensation of THP-protected indanone (**22**) with the appropriate benzaldehyde in ethanolic KOH solution, followed by removal of the THP protection.

Inhibition of Self-Mediated Aβ_{1-42} Aggregation. The ability of the indanone derivatives to inhibit $A\beta_{1-42}$ aggregation was assessed using the thioflavin T (ThT) fluorescence assay[23] with curcumin as standard. The results summarized in Table 1 showed that most indanone derivatives exhibited moderate-to-good potencies compared to that of curcumin. The optimal $A\beta_{1-42}$ aggregation inhibition potency (80.1%) was provided by compound **41** that features two hydroxy groups at the 5 and 6 positions of the indanone moiety (A ring) and one methyl-(propyl)amino substitution at the 4-position of the aldehyde (B ring). These substituted groups seem to play an important role in the inhibition of $A\beta$ aggregation. As indicated in Table 1, compound **23**, with four methoxy groups on the A and B rings, gave the lowest inhibitive activity (10.5%). Demethylation in different position in part led to a slight increase of the inhibitory activity for several compounds (14.2−28.4%). It is interesting that most of the substituted amino groups at the 4-position of the B ring (R_4) generally gave better results of the inhibitive activity 23.7−80.1%), with the exception of

Scheme 2. Synthesis of Target Compounds 23–41[a]
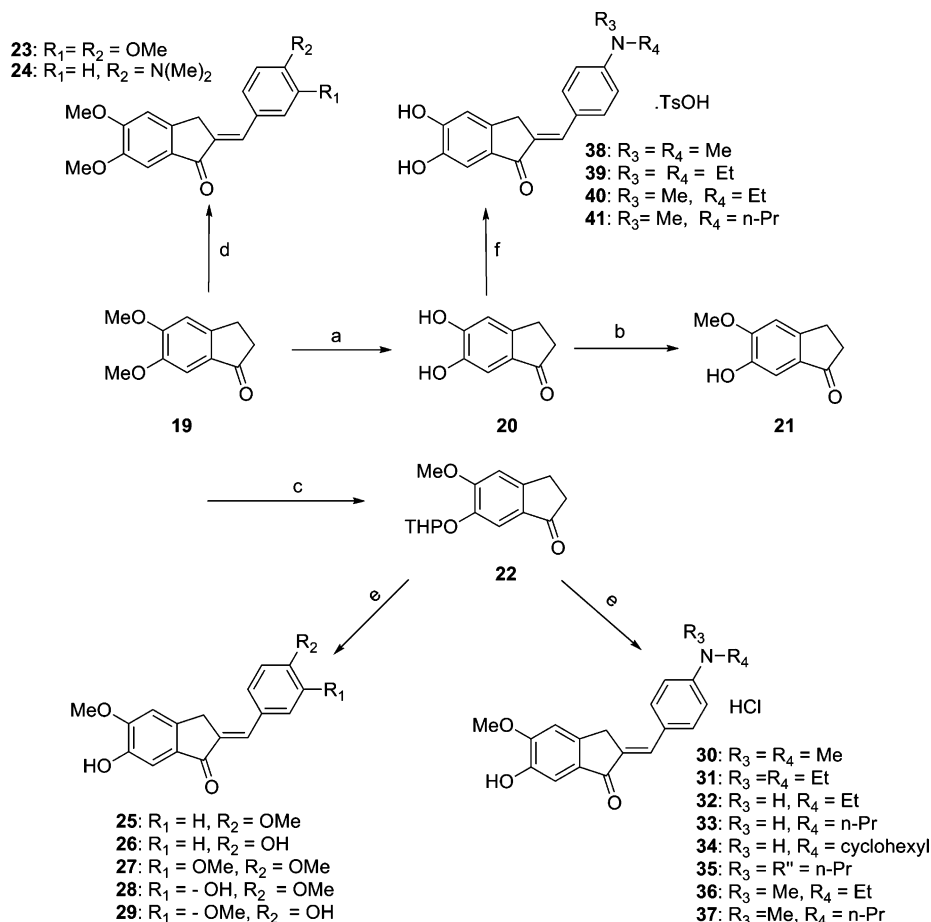
[a]Reagents and conditions: (a) BBr₃, DCM, −78 °C; (b) Li₂CO₃, MeI, DMF, 55 °C; (c) 3,4-2H-dihydropyran, PPTS, DCM, rt; (d) appropriate benzaldehyde, 4% KOH, EtOH, rt; (e) (i) appropriate benzaldehyde, 4% KOH, EtOH, rt; (ii) 2 M HCl, EA/2-butanone, reflux; (f) appropriate benzaldehyde, TsOH, reflux.

compounds **32** and **34** (NHEt, 10.5%, NH-cyc-Hex, 16.8%). Compound **24**, featuring two hydroxy groups at 5 and 6 positions of the A ring and one diethylamino group at the 4-position of the B ring, gave 23.7% inhibition activity. On the other hand, compound **30**, with a hydroxy at the 6-position of the A ring, led to 33.8% inhibition. These results imply that the methoxy group is not favorable for inhibition activity. Fixation of the A ring with a 5-methoxy-6-hydroxy substitution of the amino groups also led to changes in measured inhibitory activity (compound **31**, diethylamino group at the 4-position of the B ring, 40.7%; compound **33**, propylamino group, 38.4%, and others). Compounds **38–41**, with two hydroxy groups on the A ring, gave 61.4–80.1% inhibitory activity, which suggested that the two hydroxy groups at the 5 and 6 positions of the A ring seem to be beneficial to their activities.
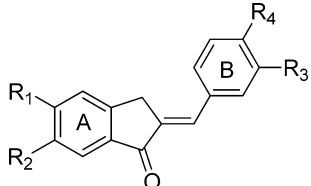
Effects on Abundance of Aβ Fibrils by Compound 41. To complement the ThT binding assay, Aβ$_{1-42}$ aggregation was also monitored by transmission electron microscopy (TEM) (Figure 2). The results showed that the sample of Aβ$_{1-42}$ alone had aggregated into amyloid fibrils after 24 h of incubation, while only small bulk aggregates were visible and no characteristic fibrils were observed in the sample of Aβ$_{1-42}$ in the presence of **41**. The TEM results were well consistent with the results of ThT measurements, which strongly proved that **41** can inhibit Aβ$_{1-42}$ fibrils formation.

Antioxidant Activity in Vitro. The antioxidant activity of the benzylideneindanone derivatives were determined by an oxygen radical absorbance capacity assay that uses fluorescein (ORAC-FL), according to the method originally described by Ou et al.[24,25] and further modified by Dávalos et al.[26] Trolox, a vitamin E analogue, was used as a standard, and the antioxidant activity was expressed as Trolox equivalents, (μmol of Trolox)/(μmol of tested compound). As shown in Table 1, most of the target compounds demonstrated good to excellent antioxidant activity with ORAC-FL values of 2.60–9.37 Trolox equivalents. Upon comparison of the ORAC-FL values of compound **23** with that of the other compounds, it is apparent that the free hydroxy or amino substituents are crucial to the radical scavenging ability.

Inhibition of MAOs in Vitro. To further study the multipotent biological profile of the target compounds, the inhibitory activity against MAO-A and MAO-B (recombinant human enzyme) was determined and compared with that of ladostigil, which was an MAO-B inhibitor approved to carry out phase II clinical trial by FDA. As shown in Table 2, most of the indanone derivatives were effective in inhibiting MAO-B in the micromolar range. Compound **41**, the most potent MAO-B inhibitor with an IC₅₀ of 7.50 μM, is about 5-fold more potent than ladostigil. However, when the hydroxy group at the 5-position of the A ring was replaced by a methoxy group (compound **37**), the inhibitory activity to both MAO-A (4.3%

Table 1. Inhibition of $A\beta_{1-42}$ Aggregation and Oxygen Radical Absorbance Capacity (ORAC, Trolox Equivalents) by Curcumin and Benzylideneindanone Derivatives 23−41

Comp.	R_1	R_2	R_3	R_4	Inhibition of $A\beta_{1-42}$ aggregation(%)[a]	ORAC[b]
Curcumin	-	-	-	-	52.1±2.7	2.57±0.17
23	OMe	OMe	OMe	OMe	10.5±0.4	0.65±0.04
24.HCl	OMe	OMe	H	N(Me)₂	23.7±0.7	7.85±0.68
25	OMe	OH	H	OMe	28.4±0.9	4.83±0.31
26	OMe	OH	H	OH	16.3±0.5	9.37±0.59
27	OMe	OH	OMe	OMe	19.5±1.0	4.55±0.26
28	OMe	OH	OH	OMe	24.0±1.3	3.55±0.20
29	OMe	OH	OMe	OH	14.2±0.7	6.58±0.55
30.HCl	OMe	OH	H	N(Me)₂	33.8±1.3	7.90±0.52
31.HCl	OMe	OH	H	N(Et)₂	40.7±1.6	3.96±0.33
32.HCl	OMe	OH	H	NHEt	10.5±0.3	4.75±0.39
33.HCl	OMe	OH	H	NHPr	38.4±1.1	4.14±0.40
34.HCl	OMe	OH	H	HN—⬡	16.8±0.5	3.60±0.25
35.HCl	OMe	OH	H	N(Pr)₂	38.0±1.5	2.75±0.11
36.HCl	OMe	OH	H	NMeEt	42.1±2.0	5.34±0.23
37.HCl	OMe	OH	H	NMePr	50.0±2.4	5.11±0.38
38.TsOH	OH	OH	H	N(Me)₂	61.4±1.9	5.63±0.44
39.TsOH	OH	OH	H	N(Et)₂	67.0±3.8	5.41±0.26
40.TsOH	OH	OH	H	NMeEt	74.4±2.9	5.87±0.41
41.TsOH	OH	OH	H	NMePr	80.1±4.0	5.60±0.59

[a]The thioflavin-T fluorescence method was used. Values are expressed as the mean ± SD from at least two independent measurements. All values were obtained at 20 μM concentration of the tested compounds. [b]The mean ± SD of the three independent experiments. Data are expressed as μmol of Trolox equivalent/μmol of tested compound.

at 50 μM) and MAO-B (IC$_{50}$ = 40.5 μM) decreased dramatically, which indicated that the hydroxy group at the 5-position of the A ring is critical to the activity.

Docking Study of Compound 41 to MAO-B. In order to evaluate the binding modes of this class of indanone derivatives with respect to human MAO-B, docking simulations were

$A\beta$ (0h) $A\beta$+ **41** (24h) $A\beta$ alone (24h)
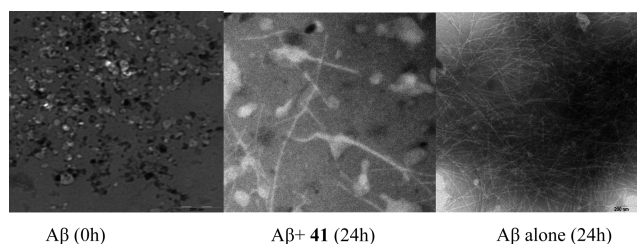
Figure 2. TEM images of $A\beta_{1-42}$ (20 μM) in the presence and absence of 20 μM compound **41** after 24 h of aggregation.

carried out using the CDOCKER program in the Discovery Studio 2.1 software based on the X-ray crystal structure of human MAO-B (PDB entry 2Z5X). On the basis of the in vitro inhibition results, we selected compound **41**, the highest MAO-B inhibitor, as a typical ligand for the evaluation. As shown in Figures 3 and 4, the indanone moiety of compound **41** was close to the FAD cofactor, adopting parallel $\pi-\pi$ interactions with Tyr398 (4.45 Å). There were two hydrogen bonds between the indanone moiety and the FAD cofactor (FAD-NH⋯41-O11; 41-OH⋯FAD-O), which could explain the critical function of hydroxy group at the 5-position of the A ring. Meanwhile, the benzene amine group interacts with many hydrophilic acids, such as Phe168, Pro 115, Ser200, Leu 187, ILe199, Thr 131, and so on.

Metal Binding Properties of Compound 41. The chelation ability of compound **41** toward biometals such as Cu(II), Fe(II), and Zn(II) was studied by UV−vis spectrometry (Figure 5). When CuSO$_4$ was added, the maximum absorption at 437 nm exhibited a red shift to 451 nm, which indicated the formation of **41**−Cu(II). However, with the addition of FeSO$_4$ and ZnCl$_2$, there was no significant shift. Interestingly, after FeSO$_4$ was added, the absorption at 437 nm decreased obviously, which indicated that there was a possible interaction between **41** and Fe(II).
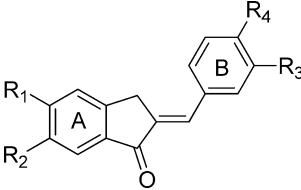
To determine the stoichiometry of the complex **41**−Cu(II), the molar ratio method was used by preparing solutions of compound **41** with increasing amounts of CuSO$_4$. The UV spectra were used to obtain the absorbance of the complex of CuSO$_4$ and **41** at different concentrations at 451 nm. According to the Figure 6, the absorbance linearly increased at first. When the mole fraction of Cu(II) to **41** was more than 0.8, the absorbance tended to be stable. Therefore, two straight lines were drawn with the intersection point at a mole fraction of 0.8, revealing a 1:1 stoichiometry for complex **41**−Cu(II).

Effects on Metal-Induced $A\beta_{1-42}$ Aggregation by Compound 41. In order to investigate the effects of **41** on Cu(II)-induced $A\beta$ aggregation, we carried out two individual studies (Figure 7 and Figure 8): inhibition activity of metal-induced $A\beta_{1-42}$ aggregation by compound **41** and disaggregation effects of **41** on Cu(II)-induced $A\beta_{1-42}$ aggregates. ThT binding assay and transmission electron microscopy (TEM) were used to identify the degree of $A\beta$ aggregation.

As shown in Figure 7, Cu(II) could accelerate the aggregation of $A\beta$. However, the rate of $A\beta$ aggregation slowed when adding **41** to the samples, which suggested that **41** could inhibit Cu(II)-induced $A\beta$ aggregation noticeably through chelating with Cu(II). The results of TME are consistent with ThT binding assay results. For the disaggregation studies (Figure 8), **41** (50 μM) was added to $A\beta$ fibrils generated by reacting $A\beta$ (50 μM) with 1 equiv of Cu(II) (50 μM) for 24 h at 37 °C with constant agitation. The TME figures illustrated

Table 2. Inhibitory Activity of Typical Molecules with Human Recombinant MAO Isoforms and Selectivity Index (SI) Values

compd	R_1	R_2	R_3	R_4	IC$_{50}$ (μM)[a]		SI[b]
					MAO-A	MAO-B	
31·HCl	OMe	OH	H	N(Et)$_2$	19.3%[c]	18.55 ± 1.1	
36·HCl	OMe	OH	H	NMeEt	14.4%[c]	28.7 ± 1.7	
37·HCl	OMe	OH	H	NMePr	4.3%[c]	40.5 ± 3.0	
38·TsOH	OH	OH	H	N(Me)$_2$	38.5 ± 2.4	10.90 ± 0.8	3.53
39·TsOH	OH	OH	H	N(Et)$_2$	41.4 ± 3.1	7.71 ± 0.5	5.36
40·TsOH	OH	OH	H	NMeEt	35.2 ± 0.9	10.7 ± 0.3	3.29
41·TsOH	OH	OH	H	NMePr	37.7 ± 2.2	7.50 ± 0.9	5.03
clorgyline					4.1 ± 0.2 nM	nt	
ladostigil					nt	37.1 ± 3.1	
selegiline					70.2 ± 3.8	18.5 ± 2.1 nM	37945
rasagiline					0.7[d]	0.014[d]	50

[a]All IC$_{50}$ values shown in this table are the mean ± SEM from three experiments. [b]hMAO-B selectivity index = IC$_{50}$(hMAO-A)/IC$_{50}$(hMAO-B). [c]The inhibition rate of compounds to MAO-A at 50 μM. [d]Reference 27.
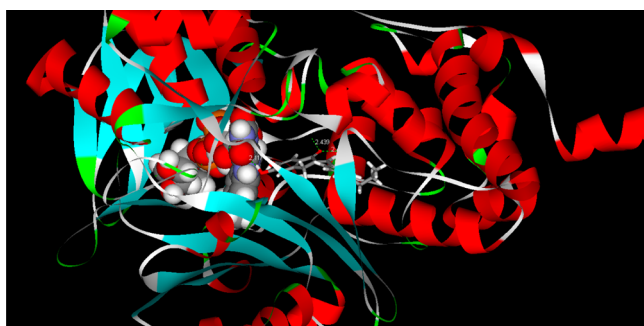

Figure 3. Predicted positions of **41** into hMAO-B catalytic sites. Compound **41** and the FAD cofactor were depicted using stick and space fill representation, respectively.
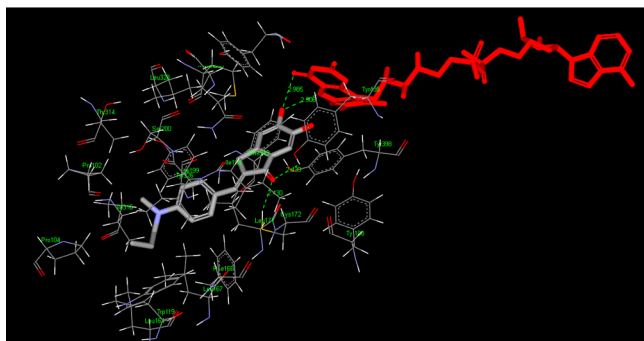

Figure 4. Representation of compound **41** docked into the binding site of MAO-B highlighting the protein residues that form the main interactions with the inhibitor. Hydrogen-bonding interactions between ligand and residues are shown with the green line.

Figure 5. UV spectra of compound **41** (50 μM) alone and in the presence of 50 μM ZnSO$_4$, CuSO$_4$, and FeSO$_4$.

compound **41** is capable of inhibiting Cu(II)-induced Aβ aggregation and disassembling the well-structured Aβ fibrils.

■ CONCLUSION

Most benzylideneindanone derivatives exhibited multifunctional activity as potential anti-AD drugs, which include significant ability to inhibit self-induced Aβ aggregation and MAO-B activity and to act as antioxidants and biometal chelators. Among the synthesized compounds, compound **41**, 5,6-dihydroxy-2-(4-(methyl(propyl)amino)benzylidene)-2,3-dihydro-1*H*-inden-1-one 4-methylbenzenesulfonate, gave the greatest inhibitory potency toward self-induced Aβ aggregation (80.1%, 20 μM), which was proved by TEM. Meanwhile, this compound was also an excellent antioxidant (ORAC-FL value of 5.60) and MAO-B inhibitor (IC$_{50}$ = 7.5 μM). UV−vis

that much fewer Aβ aggregates were observed after adding **41**, which indicated that **41** is able to alter the structure of metal-triggered Aβ aggregates, affording their disaggregation. In short, on the basis of these TEM results, we can conclude that
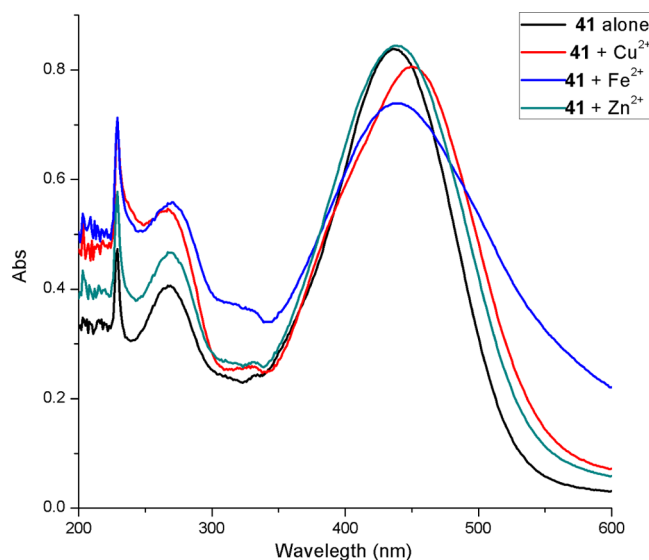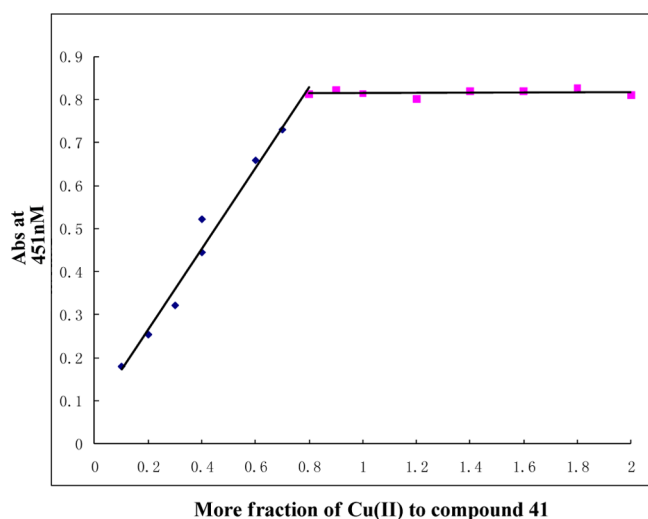
Figure 6. Determination of the stoichiometry of complex **41**−Cu(II) by molar ratio method.

spectrometry and TME results confirmed that compound **41** not only is a good biometal chelator by inhibiting Cu(II)-induced Aβ aggregation but also could disassemble the well-structured Aβ fibrils. Such multifunctional properties highlight compound **41** as an interesting candidate for further studies directed to the development of novel drugs in the treatment of AD.

■ EXPERIMENTAL SECTION

General Information. The ^{1}H NMR and ^{13}C NMR spectra were recorded using TMS as the internal standard on a Bruker BioSpin GmbH spectrometer at 400.132 and 100.614 MHz, respectively. Coupling constants are given in Hz. MS spectra were recorded on an Agilent LC−MS 6120 instrument with an ESI and APCI mass selective detector. High-resolution mass spectra were obtained using a Shimadzu LCMS-IT-TOF mass spectrometer. Flash column chromatography was performed using silica gel (200−300 mesh) purchased from Qingdao Haiyang Chemical Co. Ltd. or alumina from Sinopharm Chemical Reagent Co. Ltd. All the reactions were monitored by thin layer chromatography using silica gel. The purity of compounds **23**−**41** (higher than 95%) was confirmed through HPLC (Agilent Technologies 1200 series system, TC-C18 column (4.6 mm × 250 mm, 5 μm), and the compounds eluted with methanol/water (0.1% TFA), 75:25, at a flow rate of 0.5 mL/min).

General Procedure for the Preparation of 7−9. To a solution of **4**, **5**, or **6** (1 mmol) and PPTS (1.5 mmol) in DCM (10 mL) was added 3,4-2H-dihydropyran (1.5 mol). The mixture was stirred for 10 h and concentrated in vacuum. The crude residue was purified by flash chromatography on silica gel to furnish the oil products **7−9**.

General Procedure for the Preparation of 11−16. A mixture of 4-fluorobenzaldehyde (**10**, 2 mmol), TBAB (1 mmol), K$_2$CO$_3$ (2 mmol), and the appropriate amine (10 mmol) in DMSO was heated at 90 °C for 20 h. After cooling to room temperature, the mixture was diluted with 50 mL of water and then extracted with ethyl acetate (50 mL × 3). The combined organic phase was washed by water (10 mL), brine (10 mL), dried over Na$_2$SO$_4$, filtered, and concentrated in vacuum to give the crude product which was purified by flash chromatography on silica gel to furnish the oil products **11−16**.

General Procedure for the Preparation of 17 and 18. A mixture of **11** or **12** (1 mmol), formaldehyde (5 mmol), and formic acid (3 mmol) was refluxed for 3 h and concentrated in vacuum. The crude product was purified by flash chromatography on silica gel to furnish the oil product **17** or **18**.

5,6-Dihydroxy-2,3-dihydro-1H-inden-1-one (20). BBr$_3$ (6 mmol) was added slowly to a mixture of 5,6-dimethoxy-1-indanone (**19**, 2 mmol) in 20 mL of DCM at −78 °C. After 3 h, the mixture was cooled to room temperature and stirred for 1 h. Then water (50 mL) was added to provide a red solid, which was filtered and dried to yield **20**.
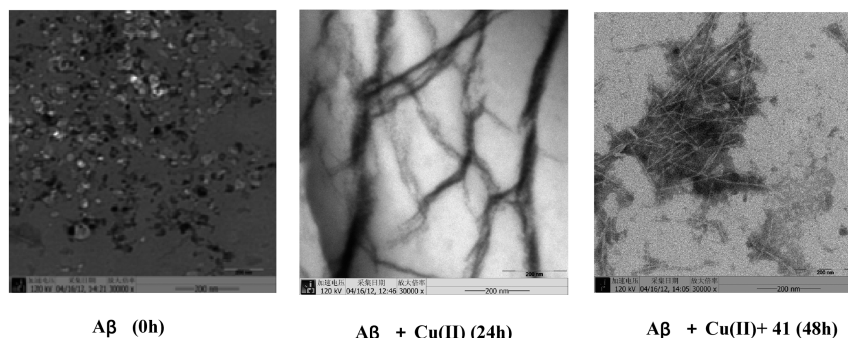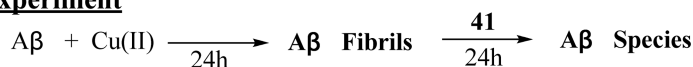
6-Hydroxy-5-methoxy-2,3-dihydro-1H-inden-1-one (21). Li$_2$CO$_3$ (2 mmol) and MeI (2.5 mmol) were added to a solution of **20** (2 mmol) in 15 mL of DMF. The mixture was stirred at 55 °C for 3 h and cooled to room temperature. Ethyl acetate (150 mL) was added to the mixture and the organic phase was washed by water (50 mL × 3), brine (10 mL), dried over Na$_2$SO$_4$, filtered, and concentrated in vacuum to furnish a yellow solid **21**.

5-Methoxy-6-(tetrahydro-2H-pyran-2-yloxy)-2,3-dihydro-1H-inden-1-one (22). 3,4-2H-Dihydropyran (1.5 mol) was added to a solution of **20** (1 mmol) and PPTS (1.5 mmol) in DCM (10 mL).
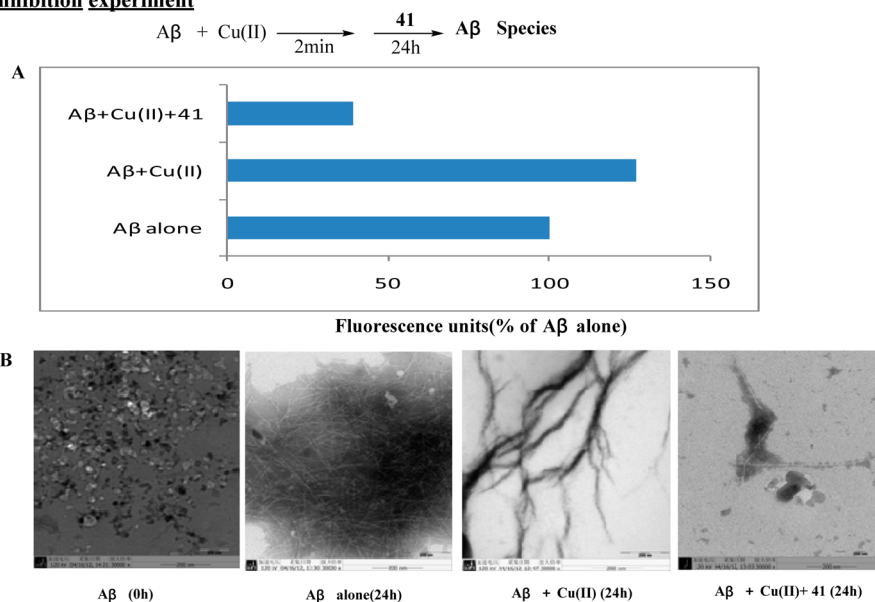
Figure 7. Visualization of Aβ species from inhibition experiments: (top) scheme of the inhibition experiment; (A) results of ThT binding assay; (B) TEM images of samples.

Disaggregation experiment

$$A\beta\ +\ Cu(II)\ \xrightarrow{\quad 24h\quad}\ A\beta\ \ Fibrils\ \xrightarrow[\ 24h\ ]{\ 41\ }\ A\beta\ \ Species$$

Aβ (0h)   **Aβ + Cu(II) (24h)**   **Aβ + Cu(II)+ 41 (48h)**

Figure 8. Visualization of Aβ species from disaggregation experiments: (top) scheme of the disaggregation experiment; (bottom) TEM images of samples.

The mixture was stirred for 10 h and concentrated in vacuum. The crude product was purified by flash chromatography on a silica gel to furnish the oil product **22**.

General Procedure for the Preparation of 23 and 24. An amount of 5 mL of 4% KOH was added to a solution of 5,6-dimethoxy-1-indanone (**19**) (384 mg, 2 mmol) with the appropriate benzaldehyde (2 mmol) in 5 mL of EtOH. After the mixture was stirred at room temperature for 2−5 h, the solid was filtered, washed with water, and crystallized from methanol.

2-(3,4-Dimethoxybenzylidene)-5,6-dimethoxy-2,3-dihydro-1H-inden-1-one (23). 5,6-Dimethoxy-1-indanone (**19**) was treated with 3,4-dimethoxybenzaldehyde according to the general procedure to give the desired product **23** as a yellow solid, 83% yield. Mp = 195.6−196.3 °C; ^{1}H NMR (400 MHz, CDCl$_3$) δ 7.54 (s, 1H), 7.34 (s, 1H), 7.30−7.25 (m, 1H), 7.17 (d, J = 1.7 Hz, 1H), 6.99 (s, 1H), 6.94 (d, J = 8.4 Hz, 1H), 4.00 (s, 2H), 3.97−3.93 (m, 12H); ^{13}C NMR (101 MHz, CDCl$_3$) δ 194.37, 155.26, 149.64, 144.52, 133.36, 132.52, 131.26, 128.63, 124.37, 113.34, 111.32, 107.21, 105.10, 56.26, 55.97, 32.06; LCMS (APCI) m/z [(M + H)]$^+$ = 341.4; HRMS (ESI) m/z calcd for C$_{20}$H$_{20}$O$_5$, 341.1384; found, 341.1378. Purity: 95.1% (by HPLC).

2-(4-(Dimethylamino)benzylidene)-5,6-dimethoxy-2,3-dihydro-1H-inden-1-one (24). 5,6-Dimethoxy-1-indanone (**19**) was treated with 4-(dimethylamino)benzaldehyde according to the general procedure to give the desired product **24** as a yellow solid, 85% yield. Mp = 205.0−205.9 °C; ^{1}H NMR (400 MHz, CDCl$_3$) δ 7.57 (m, 3H), 7.34 (s, 1H), 6.97 (s, 1H), 6.73 (d, J = 8.8 Hz, 2H), 3.99 (s, 3H), 3.94 (s, 3H), 3.90 (s, 2H), 3.04 (s, 6H). ^{13}C NMR (101 MHz, CDCl$_3$) δ 193.29, 154.81, 150.98, 149.45, 144.36, 133.35, 132.45, 131.71, 130.67, 123.44, 111.96, 107.23, 105.04, 56.21, 56.14, 40.08, 32.37; LCMS (APCI) m/z [(M + H)]$^+$ = 324.4; HRMS (ESI) m/z calcd for C$_{20}$H$_{21}$NO$_3$, 324.1594; found, 324.1588. Purity: 99.7% (by HPLC).

General Procedure for the Preparation of 25−37. To a solution of the THP protected indanone (**22**) (261 mg, 1 mmol) in EtOH (3 mL), the appropriate benzaldehyde (1 mmol) and 3 mL of 4% KOH were added. After the mixture was stirred at room temperature for 2−5 h, the produced light yellow solid was filtered. The solid was dissolved in 10 mL of EA/butanone (1:1) and treated with 5 mL of 2 M HCl. The mixture was refluxed for 2 h and then concentrated in vacuum. The solid residue was washed with water and crystallized from methanol.

6-Hydroxy-5-methoxy-2-(4-methoxybenzylidene)-2,3-dihydro-1H-inden-1-one (25). Intermediate **22** was treated with 4-methoxybenzaldehyde according to the general procedure to give the desired product **25** as a yellow solid, yield 77%. Mp = 227.0−227.6 °C; ^{1}H NMR (400 MHz, DMSO-d$_6$) δ 9.48 (s, 1H), 7.70 (d, J = 8.7 Hz, 2H), 7.36 (s, 1H), 7.16 (s, 1H), 7.12−6.99 (m, 3H), 3.93 (s, 2H), 3.91 (s, 3H), 3.83 (s, 3H); ^{13}C NMR (101 MHz, DMSO-d$_6$) δ 191.96, 160.25, 154.39, 146.89, 143.32, 133.60, 132.25, 130.77, 130.36, 127.70,

114.49, 108.22, 108.05, 55.85, 55.30, 31.46; LCMS (APCI) m/z [(M − H)]$^-$ = 295.3; HRMS (ESI) m/z calcd for C$_{18}$H$_{16}$O$_4$, 297.1121; found, 297.1119. Purity: 95.2% (by HPLC).

6-Hydroxy-2-(4-hydroxybenzylidene)-5-methoxy-2,3-dihydro-1H-inden-1-one (26). Intermediate **22** was treated with compounds **7** according to the general procedure to give the desired product **26** as a yellow solid, yield 65%. Mp = 278.8−279.5 °C; ^{1}H NMR (400 MHz, DMSO-d$_6$) δ 10.07 (s, 1H), 9.52 (s, 1H), 7.58 (d, J = 8.6 Hz, 2H), 7.31 (s, 1H), 7.14 (s, 1H), 7.06 (d, J = 2.4 Hz, 1H), 6.92−6.79 (m, 2H), 3.89 (s, 5H). ^{13}C NMR (101 MHz, DMSO-d$_6$) δ 191.99, 158.75, 154.18, 146.69, 143.23, 132.51, 131.25, 130.44, 126.23, 115.92, 115.82, 108.20, 107.93, 99.39, 55.82, 31.51. LCMS (APCI) m/z [(M − H)]$^-$ = 281.3; HRMS (ESI) m/z calcd for C$_{17}$H$_{14}$O$_4$, 283.0965; found, 283.0964. Purity: 98.9% (by HPLC).

2-(3,4-Dimethoxybenzylidene)-6-hydroxy-5-methoxy-2,3-dihydro-1H-inden-1-one (27). Intermediate **22** was treated with 3,4-dimethoxybenzaldehyde according to the general procedure to give the desired product **27** as a yellow solid, yield 63%. Mp = 177.8−178.4 °C; ^{1}H NMR (400 MHz, DMSO-d$_6$) δ 7.35 (s, 1H), 7.30 (d, J = 7.0 Hz, 2H), 7.17 (s, 1H), 7.06 (d, J = 10.9 Hz, 2H), 3.96 (s, 2H), 3.90 (s, 3H), 3.84 (s, 3H), 3.82 (s, 3H); ^{13}C NMR (101 MHz, DMSO-d$_6$) δ 192.44, 150.60, 149.26, 143.86, 134.17, 131.72, 130.85, 128.40, 124.88, 113.83, 112.36, 108.75, 56.36, 56.05, 55.95, 31.87; LCMS (APCI) m/z [(M − H)]$^-$ = 325.4; HRMS (ESI) m/z calcd for C$_{19}$H$_{18}$O$_5$, 327.1227; found, 327.1224. Purity: 98.7% (by HPLC).

6-Hydroxy-2-(3-hydroxy-4-methoxybenzylidene)-5-methoxy-2,3-dihydro-1H-inden-1-one (28). Intermediate **22** was treated with compounds **8** according to the general procedure to give the desired product **28** as a yellow solid, yield 62%. Mp = 216.2−218.2 °C; ^{1}H NMR (400 MHz, DMSO-d$_6$) δ 7.30 (d, J = 13.3 Hz, 2H), 7.23−7.11 (m, 2H), 7.07 (s, 1H), 6.88 (d, J = 7.8 Hz, 1H), 3.92 (s, 2H), 3.88 (s, 3H), 3.85 (s, 3H); ^{13}C NMR (101 MHz, DMSO-d$_6$) δ 191.97, 154.22, 148.38, 147.70, 146.76, 143.23, 132.74, 131.62, 130.44, 126.65, 124.71, 115.80, 114.13, 108.22, 107.99, 55.86, 55.58, 31.43; LCMS (APCI) m/z [(M − H)]$^-$ = 311.3; HRMS (ESI) m/z calcd for C$_{18}$H$_{16}$O$_5$, 313.1071; found, 313.1070. Purity: 98.9% (by HPLC).

6-Hydroxy-2-(4-hydroxy-3-methoxybenzylidene)-5-methoxy-2,3-dihydro-1H-inden-1-one (29). Intermediate **22** was treated with compounds **9** according to the general procedure to give the desired product **29** as a yellow solid, yield 60%. Mp = 222.9−223.5 °C; ^{1}H NMR (400 MHz, DMSO-d$_6$) δ 7.25 (s, 1H), 7.23−7.11 (m, 3H), 7.07 (s, 1H), 7.01 (d, J = 8.3 Hz, 1H), 3.89 (s, 3H), 3.87 (s, 2H), 3.82 (s, 3H); ^{13}C NMR (101 MHz, DMSO-d$_6$) δ 191.93, 154.30, 149.15, 146.81, 146.51, 143.22, 133.40, 131.25, 130.39, 127.97, 123.33, 116.75, 112.17, 108.21, 108.00, 55.83, 55.58, 31.48; LCMS (APCI) m/z [(M − H)]$^-$ = 311.3; HRMS (ESI) m/z calcd for C$_{18}$H$_{16}$O$_5$, 313.1071; found, 313.1074. Purity: 98.5% (by HPLC).

2-(4-(Dimethylamino)benzylidene)-6-hydroxy-5-methoxy-2,3-dihydro-1H-inden-1-one Hydrochloride (30·HCl). Intermedi-

ate **22** was treated with 4-(dimethylamino)benzaldehyde according to the general procedure to give the desired product **30** as a yellow solid, yield 65%. Mp = 217.8−219.0 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 9.46 (s, 1H), 7.56 (d, J = 8.9 Hz, 2H), 7.30 (s, 1H), 7.15 (s, 1H), 7.05 (d, J = 2.3 Hz, 1H), 6.78 (d, J = 8.9 Hz, 2H), 3.89 (s, 3H), 3.86 (s, 2H), 2.99 (s, 6H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.84, 153.88, 150.88, 146.60, 142.83, 132.21, 131.96, 130.81, 130.62, 122.44, 111.96, 108.24, 107.90, 99.48, 55.81, 31.70; LCMS (APCI) m/z [(M − H)]$^-$ = 308.4; HRMS (ESI) m/z calcd for $C_{19}H_{19}NO_3$, 310.1438; found, 310.1440. Purity: 96.2% (by HPLC).

2-(4-(Diethylamino)benzylidene)-6-hydroxy-5-methoxy-2,3-dihydro-1*H*-inden-1-one Hydrochloride (31·HCl). Intermediate **22** was treated with 4-(diethylamino)benzaldehyde according to the general procedure to give the desired product **31** as a yellow solid, yield 83%. Mp = 150.6−151.2 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.70 (s, 2H), 7.34 (s, 1H), 7.21−7.18 (m, 2H), 7.14 (s, 1H), 7.08 (s, 1H), 3.91 (s, 2H), 3.89 (s, 3H), 3.49−3.47 (m, 4H), 1.08 (t, J = 6.9 Hz, 6H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.82, 154.37, 146.86, 143.26, 132.24, 130.43, 108.21, 108.05, 55.87, 47.54, 31.50, 11.19; LCMS (APCI) m/z [(M − H)]$^-$ = 336.4; HRMS (ESI) m/z calcd for $C_{21}H_{23}NO_3$, 338.1751; found, 338.1751. Purity: 99.5% (by HPLC).

2-(4-(Ethylamino)benzylidene)-6-hydroxy-5-methoxy-2,3-dihydro-1*H*-inden-1-one Hydrochloride (32·HCl). Intermediate **22** was treated with 4-(ethylamino)benzaldehyde according to the general procedure to give the desired product **32** as a yellow solid, yield 72%. Mp = 237.0−237.7 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.61 (d, J = 8.3 Hz, 2H), 7.31 (s, 1H), 7.14 (s, 1H), 7.08 (s, 1H), 6.95 (d, J = 7.7 Hz, 2H), 3.90−3.85 (m, 5H), 3.19−3.10 (m, 2H), 1.21 (t, J = 7.1 Hz, 3H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.74, 154.34, 146.95, 143.13, 132.01, 130.47, 108.17, 108.10, 55.85, 31.52, 12.48; LCMS (APCI) m/z [(M − H)]$^-$ = 308.4; HRMS (ESI) m/z calcd for $C_{19}H_{19}NO_3$, 310.1438; found, 310.1431. Purity: 99.8% (by HPLC).

6-Hydroxy-5-methoxy-2-(4-(propylamino)benzylidene)-2,3-dihydro-1*H*-inden-1-one Hydrochloride (33·HCl). Intermediate **22** was treated with 4-(propylamino)benzaldehyde according to the general procedure to give the desired product **33** as a yellow solid, yield 75%. Mp = 222.1−222.9 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.58 (d, J = 8.2 Hz, 2H), 7.30 (s, 1H), 7.14 (s, 1H), 7.09 (s, 1H), 6.91 (d, J = 7.6 Hz, 2H), 3.90 (s, 3H), 3.89 (s, 2H), 3.10 (t, J = 7.1 Hz, 2H), 1.61 (dd, J = 14.2, 7.2 Hz, 2H), 0.95 (t, J = 7.3 Hz, 3H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.75, 154.25, 146.91, 143.04, 132.08, 130.98, 130.53, 108.18, 108.09, 55.87, 31.56, 20.50, 11.23; LCMS (APCI) m/z [(M − H)]$^-$ = 322.4; HRMS (ESI) m/z calcd for $C_{20}H_{21}NO_3$, 324.1594; found, 324.1596. Purity: 99.3% (by HPLC).

2-(4-(Cyclohexylamino)benzylidene)-6-hydroxy-5-methoxy-2,3-dihydro-1*H*-inden-1-one Hydrochloride (34·HCl). Intermediate **22** was treated with 4-(cyclohexylamino)benzaldehyde according to the general procedure to give the desired product **34** as a yellow solid, yield 77%. Mp = 250.4−251.0 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 9.57 (s, 1H), 7.46 (d, J = 8.5 Hz, 2H), 7.26 (s, 1H), 7.15 (t, J = 13.5 Hz, 2H), 6.67 (d, J = 8.5 Hz, 2H), 6.27 (d, J = 7.9 Hz, 1H), 3.89 (s, 3H), 3.85 (s, 2H), 1.92 (d, J = 9.9 Hz, 2H), 1.73 (d, J = 13.0 Hz, 2H), 1.60 (d, J = 12.0 Hz, 1H), 1.51−0.91 (m, 6H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.79, 154.21, 149.21, 146.69, 142.41, 132.51, 132.51, 130.94, 129.61, 122.11, 112.61, 112.22, 108.20, 108.02, 99.22, 55.69, 50.12, 32.39, 32.13, 31.85, 25.32, 24.40, 24.25; LCMS (APCI) m/z [(M − H)]$^-$ = 362.4; HRMS (ESI) m/z calcd for $C_{23}H_{25}NO_3$, 364.1907; found, 364.1900. Purity: 95.5% (by HPLC).

2-(4-(Dipropylamino)benzylidene)-6-hydroxy-5-methoxy-2,3-dihydro-1*H*-inden-1-one Hydrochloride (35·HCl). Intermediate **22** was treated with 4-(dipropylamino)benzaldehyde according to the general procedure to give the desired product **35** as a yellow solid, yield 80%. Mp = 122.9−123.0 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.60 (s, 2H), 7.30 (s, 1H), 7.13 (s, 1H), 7.07 (s, 1H), 7.00−6.79 (m, 2H), 3.89 (s, 3H), 3.88 (s, 2H), 3.34 (t, J = 7.5 Hz, 4H), 1.67−1.36 (m, 4H), 0.88 (t, J = 7.3 Hz, 6H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.73, 154.20, 146.91, 142.95, 132.24, 130.62, 108.19, 108.09, 55.85, 54.85, 31.57, 19.28, 10.92; LCMS (APCI) m/z [(M − H)]$^-$ = 364.5; HRMS (ESI) m/z calcd for $C_{23}H_{27}NO_3$, 366.2064; found, 366.2078. Purity: 98.9% (by HPLC).

2-(4-(Ethyl(methyl)amino)benzylidene)-6-hydroxy-5-methoxy-2,3-dihydro-1*H*-inden-1-one Hydrochloride (36·HCl). Intermediate **22** was treated with 4-(ethyl(methyl)amino)benzaldehyde according to the general procedure to give the desired product **36** as a yellow solid, yield 77%. Mp = 193.7−194.4 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.61 (d, J = 8.4 Hz, 2H), 7.32 (s, 1H), 7.15 (s, 1H), 7.07 (s, 1H), 6.93 (s, 2H), 3.92−3.87 (m, 5H), 3.49 (dd, J = 13.9, 6.8 Hz, 2H), 2.99 (s, 3H), 1.08 (t, J = 6.9 Hz, 3H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.74, 154.30, 146.94, 143.07, 132.14, 130.76, 130.54, 108.21, 108.10, 55.86, 31.54, 10.64; LCMS (APCI) m/z [(M − H)]$^-$ = 322.4; HRMS (ESI) m/z calcd for $C_{20}H_{21}NO_3$, 324.1594; found, 324.1582. Purity: 98.9% (by HPLC).

6-Hydroxy-5-methoxy-2-(4-(methyl(propyl)amino)-benzylidene)-2,3-dihydro-1*H*-inden-1-one Hydrochloride (37·HCl). Intermediate **22** was treated with 4-(methyl(propyl)amino)-benzaldehyde according to the general procedure to give the desired product **37** as a yellow solid, yield 77%. Mp = 133.1−133.8 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.57 (d, J = 8.7 Hz, 2H), 7.30 (s, 1H), 7.15 (s, 1H), 7.06 (s, 1H), 6.81 (d, J = 8.2 Hz, 2H), 3.90 (s, 3H), 3.88 (s, 2H), 3.40−3.34 (m, 2H), 2.99 (s, 3H), 1.56 (dd, J = 14.5, 7.4 Hz, 2H), 0.89 (t, J = 7.3 Hz, 3H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 192.12, 153.84, 147.28, 142.71, 132.24, 130.59, 108.22, 108.06, 55.83, 31.62, 19.20, 11.02; LCMS (APCI) m/z [(M − H)]$^-$ = 336.4; HRMS (ESI) m/z calcd for $C_{21}H_{23}NO_3$, 338.1751; found, 338.1760. Purity: 98.3% (by HPLC).

General Procedure for the Preparation of 38−41. TsOH (380 mg, 2 mmol) was added to a suspension of 5,6-dihydroxyindanone (**20**) (2 mmol) and the appropriate benzaldehyde (2 mmol) in 10 mL of toluene. After being refluxed for 10 h, the solution was cooled to room temperature. The solid was filtered off, washed with water, and crystallized from methanol.

2-(4-(Dimethylamino)benzylidene)-5,6-dihydroxy-2,3-dihydro-1*H*-inden-1-one 4-Methylbenzenesulfonate (38). 5,6-Dihydroxyindanone (**20**) was treated with 4-(dimethylamino)benzaldehyde according to the general procedure to give the desired product **38** as a yellow solid, yield 71%. Mp = 247.6−248.1 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.56−7.50 (m, 4H), 7.27 (s, 1H), 7.13 (d, J = 7.7 Hz, 2H), 7.04 (s, 1H), 6.94 (s, 1H), 6.78 (d, J = 8.9 Hz, 2H), 3.81 (s, 2H), 2.99 (s, 6H), 2.28 (s, 3H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.58, 152.67, 150.80, 145.76, 143.01, 135.74, 132.07, 131.36, 131.07, 130.55, 129.90, 128.97, 125.35, 122.65, 111.97, 108.44, 41.59, 31.42, 20.50; LCMS (APCI) m/z [(M − H)]$^-$ = 294.3; HRMS (ESI) m/z calcd for $C_{18}H_{17}NO_3$, 296.1281; found, 296.1276. Purity: 99.7% (by HPLC).

2-(4-(Diethylamino)benzylidene)-5,6-dihydroxy-2,3-dihydro-1*H*-inden-1-one 4-Methylbenzenesulfonate (39). 5,6-Dihydroxyindanone (**20**) was treated with 4-(diethylamino)benzaldehyde according to the general procedure to give the desired product **39** as a yellow solid, yield 67%. Mp = 240.1−240.7 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 10.03 (s, 1H), 9.43 (s, 1H), 7.53−7.48 (m, 4H), 7.25 (s, 1H), 7.14 (d, J = 7.6 Hz, 2H), 7.05 (s, 1H), 6.94 (s, 1H), 6.73 (d, J = 8.9 Hz, 2H), 3.81 (s, 2H), 3.41 (dd, J = 14.0, 7.0 Hz, 4H), 2.32 (s, 3H), 1.12 (t, J = 7.0 Hz, 6H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.64, 152.52, 148.19, 145.73, 142.93, 135.52, 132.47, 131.53, 130.89, 130.45, 130.03, 128.56, 126.17, 121.78, 111.71, 111.32, 108.45, 43.70, 31.45, 20.28, 12.39; LCMS (APCI) m/z [(M − H)]$^-$ = 322.4; HRMS (ESI) m/z calcd for $C_{20}H_{21}NO_3$, 324.1594; found, 324.1595. Purity: 99.5% (by HPLC).

2-(4-(Ethyl(methyl)amino)benzylidene)-5,6-dihydroxy-2,3-dihydro-1*H*-inden-1-one 4-Methylbenzenesulfonate (40). 5,6-Dihydroxyindanone (**20**) was treated with 4-(ethyl(methyl)amino)-benzaldehyde according to the general procedure to give the desired product **40** as a yellow solid, yield 66%. Mp = 210.7−211.1 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 7.54−7.51 (m, 4H), 7.27 (s, 1H), 7.10 (d, J = 7.8 Hz, 2H), 7.05 (s, 1H), 6.94 (s, 1H), 6.86 (d, J = 7.3 Hz, 2H), 3.82 (s, 2H), 3.47 (dd, J = 13.8, 6.8 Hz, 2H), 2.98 (s, 3H), 2.29 (s, 3H), 1.08 (t, J = 6.9 Hz, 3H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.53, 152.92, 145.88, 145.11, 143.23, 137.99, 132.13, 129.75, 128.12, 125.51, 111.71, 108.57, 40.15, 31.32, 20.73, 10.78; LCMS (APCI) m/z

$[(M - H)]^- = 308.4$; HRMS (ESI) m/z calcd for $C_{19}H_{19}NO_3$, 310.1438; found, 310.1426. Purity: 99.0% (by HPLC).

5,6-Dihydroxy-2-(4-(methyl(propyl)amino)benzylidene)-2,3-dihydro-1H-inden-1-one 4-Methylbenzenesulfonate (41). 5,6-Dihydroxyindanone (**20**) was treated with 4-(methyl(propyl)amino)-benzaldehyde according to the general procedure to give the desired product **41** as a yellow solid, yield 60%. Mp = 222.9–223.1 °C; ^{1}H NMR (400 MHz, DMSO-d_6) δ 9.98 (s, 1H), 9.35 (s, 1H), 7.53–7.49 (m, 4H), 7.26 (s, 1H), 7.13 (d, $J = 7.7$ Hz, 2H), 7.05 (s, 1H), 6.94 (s, 1H), 6.76 (d, $J = 8.5$ Hz, 2H), 3.81 (s, 2H), 3.40–3.32 (m, 2H), 2.97 (s, 3H), 2.30 (s, 3H), 1.56 (dd, $J = 14.5, 7.2$ Hz, 2H), 0.89 (t, $J = 7.3$ Hz, 3H); ^{13}C NMR (101 MHz, DMSO-d_6) δ 191.62, 152.57, 149.70, 145.75, 145.16, 142.94, 137.92, 132.25, 131.46, 130.70, 129.97, 128.09, 125.51, 122.12, 111.70, 111.61, 108.46, 52.96, 31.44, 20.73, 19.50, 11.13; LCMS (APCI) m/z $[(M - H)]^- = 322.2$. HRMS (ESI) m/z calcd for $C_{20}H_{21}NO_3$, 324.1594; found, 324.1594. Purity: 97.7% (by HPLC).

Inhibition of Aβ_{1-42} Peptide Aggregation.[23] Hexafluoro-2-propanol (HFIP) pretreated Aβ_{1-42} samples (Millipore) were dissolved in a 50 mM phosphate buffer (pH 7.4) in order to have a stable stock solution ($[A\beta] = 200$ μM). The peptide was incubated in 50 mM phosphate buffer (pH 7.4) at 37 °C for 48 h (final Aβ concentration of 50 μM) with or without the tested compound at 20 μM. After incubation, the samples were diluted to a final volume of 200 μL with 50 mM glycine–NaOH buffer (pH 8.0) containing thioflavin T. Then a 300 s time scan of fluorescence intensity was performed ($\lambda_{exc} = 450$ nm; $\lambda_{em} = 485$ nm), and values at plateau were averaged after subtracting the background fluorescence of the thioflavin T solution.

Antioxidant Activity Assay. The antioxidant activity was determined by the oxygen radical absorbance capacity fluorescein (ORAC-FL) assay.[24−26] All the assays were conducted with 75 mM phosphate bufffer (pH 7.4), and the final reaction mixture was 200 μL. Antioxidant (20 μL) and fluorescein (120 μL, 300 nM final concentration) were placed in the wells of a black 96-well plate, and the mixture was incubated for 10 min at 37 °C. Then AAPH (Aldrich) solution (60 μL, 12 mM final concentration) was added rapidly. The plate was immediately placed into a Spectrafluor Plus plate reader (Tecan, Crailsheim, Germany), and the fluorescence was measured every 60 s for 4 h with excitation at 485 nm and emission at 520 nm. Trolox was used as standard (1–10 μM, final concentration). A blank (FL + AAPH) using phosphate buffer instead of antioxidant and Trolox calibration were carried out in each assay. The samples were measured at different concentrations (0.5–10 μM). All reaction mixtures were prepared 4-fold, and at least four independent runs were performed for each sample. Fluorescence measurements were normalized to the curve of the blank (without antioxidant). The ORAC-FL values were calculated as described in the reference, and the final results were in (μM Trolox equivalent)/(μM pure compound).

Effects of 41 on Aβ_{1-42} Aggregation by TEM Study.[28] Aβ_{1-42} peptide (Millipore) was dissolved in 10 mM phosphate buffer (pH 7.4) at 4 °C to give an 80 μM solution. Aβ_{1-42} was incubated in the presence and absence of **41** at 37 °C. The final concentrations of both Aβ_{1-42} and **41** were 20 μM. At specified time points, aliquots of 10 μL samples were placed on a carbon-coated copper/rhodium grid. After 1 min, the grid was washed with water and negatively stained with 2% phosphomolybdic acid solution for 1 min. After the excess of staining solution was drained off by means of a filter paper, the specimen was transferred for examination in a transmission electron microscope (JEOL JEM-1400).

Inhibition of MAO Activity.[29] The potential effects of the test drugs on hMAO activity were investigated by measuring their effects on the production of H_2O_2 from p-tyramine, using the Amplex Red MAO assay kit (Molecular Probes, Inc.) and recombinant human MAO-A or MAO-B (Sigma-Aldrich) according to published procedures.

Docking Study. The simulation system was built based on the X-ray crystal structure of MAO-B which was obtained from the Protein Data Bank (PDB entry 2Z5X). The original ligand was removed, while water molecules present in the PDB file were maintained in their positions. The 3D structures of **41** were generated and optimized by the Discovery Studio 2.1 package (Accelrys Inc., San Diego, CA). The CDOCKER program of the Discovery Studio 2.1 software was used to perform docking simulations, which allows full flexibility of the ligands.

Metal Chelation.[30] Compound **41** was tested as a metal chelator, using difference UV–vis spectra recorded in methanol at 298 K with wavelength ranging from 200 to 600 nm. Numerical subtraction of the spectra of the metal alone and the compound alone from the spectra of the mixture gave the difference UV–vis spectra due to complex formation. A fixed amount of **41** (50 μM) was mixed with growing amounts of copper ion (5–100 μM), and the difference UV–vis spectra were examined to investigate the ratio of ligand/metal in the complex.

Effects of 41 on Metal-Induced Aβ_{1-42} Aggregation and Disaggregation Experiments by ThT Method and TME.[28] HEPES buffer solutions (20 μM, pH 6.6) containing 150 μM NaCl were prepared with distilled water. Solutions of Cu^{2+} were prepared from standards to final concentrations of 200 μM using the HEPES buffer at pH 6.6. Solutions of **41** were prepared in DMSO in 10 mM for storage and diluted with HEPES buffer before use. To study the effects of **41** on the metal-induced Aβ_{1-42} aggregation, Aβ_{1-42} (50 μM) was incubated with 50 μM Cu^{2+} in HEPES buffer at pH 6.6, without or with compound **41** (50 μM). The incubation was performed at 37 °C for 24 h. After incubation, the samples were diluted to a final volume of 180 μL with 50 mM glycine–NaOH buffer (pH 8.5) containing 5 μM thioflavin T. Fluorescence was measured at 450 nm (k_{ex}) and 485 nm (k_{em}) using a monochromator based multimode microplate reader (Infinite M1000). The TEM study was carried out as in the previous procedure.[28]

Disaggregation experiments were carried out according to published procedures.[31,32]

ASSOCIATED CONTENT

Ⓢ Supporting Information
HPLC chromatograms of compounds **23−41**. This material is available free of charge via the Internet at http://pubs.acs.org.

Accession Codes
PDB code for structure of human MAO-B is 2Z5X.

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Notes
The authors declare no competing financial interest.

ABBREVIATIONS USED

AD, Alzheimer's disease; Aβ, β-amyloid peptide; MAO-B, monoamine oxidase B; MAO-A, monoamine oxidase A; ACh, acetylcholine; FDA, Food and Drug Administration; CNS, central nervous system; ROS, reactive oxygen species; MTDL, multitarget-directed ligand; AChE, acetylcholinesterase; TBAB, tetrabutylammonium bromide; ThT, thioflavin T; TEM, transmission electron microscopy; FAD, flavin adenine dinucleotide; PDB, Protein Data Bank; SAR, structure–activity relationship

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